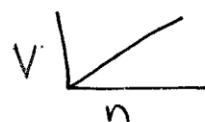


Avagadro's Law

Relates Volume and amount (n)
Keeping T and P constant

Direct $\uparrow V$, $\uparrow n$ 

n = amount in moles

Formula

$$\frac{V_1}{n_1} = \frac{V_2}{n_2}$$

↑
use when
missing V_2

or

$$\frac{n_1}{V_1} = \frac{n_2}{V_2}$$

↑
use when
missing n_2